

# Electrical Characteristic of Fe<sub>2</sub>O<sub>3</sub> Thick Film Ceramics Made from Local Mineral in Air and Ethanol Atmosphere

Endi SUHENDI<sup>1\*</sup>, Hera NOVIA<sup>1</sup> and Dani Gustaman SYARIF<sup>2</sup>

<sup>1</sup>Physics Department, UPI, Jl. Dr. Setiabudhi, 225 Bandung, Indonesia.

<sup>2</sup>PTNBR-BATAN, Jl. Tamansari 71, Bandung 40132, Indonesia.

\*endis@upi.edu

**ABSTRACT.** In order to find possibility to utilize local mineral for high value products, fabrication of Fe<sub>2</sub>O<sub>3</sub> thick film ceramics from local mineral of yarosite has been carried out and characteristics especially electrical characteristics of the thick film ceramics have been studied. Powder Fe<sub>2</sub>O<sub>3</sub> was derived from yarosite using precipitation method. The nano powder of Fe<sub>2</sub>O<sub>3</sub> was mixed with organic vehicle (OV) consists of alpha terpineol and ethyl cellulose to form a paste. The paste was screen printed on alumina substrates using screen sizes of 183 mesh. The green thick films were fired at temperature of 950°C for 90 minutes. The fired film was analyzed using an x-ray diffractometer (XRD) to know its crystal structure and a Scanning Electron Microscope (SEM) to know its morphology. Electrical resistance of the thick films was measured at various temperatures in air and air containing ethanol gas.

**Key words:** Ceramic, Fe<sub>2</sub>O<sub>3</sub>, Thick film, ethanol sensor

## 1. INTRODUCTION

In order to find possibility to utilize local mineral for high value products, study of fabrication of a special product such as gas sensor has to be carried out. Here, fabrication of Fe<sub>2</sub>O<sub>3</sub> thick film ceramics from local mineral of yarosite has been done and their electrical characteristics has been studied.

## 2. MATERIALS AND METHOD

Powder Fe<sub>2</sub>O<sub>3</sub> was derived from yarosite using precipitation method. Mineral of yarosite was dissolved in HCl. Certain amount of NH<sub>4</sub>OH was added into the solution until forming precipitation. The precipitate was dried and calcined at 700°C for 2 hours. The powder was analyzed using an x-ray diffractometer (XRD). The nano powder of Fe<sub>2</sub>O<sub>3</sub> was mixed with organic vehicle (OV) consists of alpha terpineol and 10 weight % ethyl cellulose to form a paste. The paste was screen printed on alumina substrates using screen sizes of 183 mesh. The green thick films were fired at temperature of 950°C for 90 minutes. The fired film was analyzed using XRD to know its crystal structure and a Scanning Electron Microscope (SEM) to know its morphology. Electrical resistance of the thick films was measured at various temperatures in air and air containing ethanol gas.

## 3. RESULTS AND DISCUSSION

Visual appearance of Fe<sub>2</sub>O<sub>3</sub> powder derived from yarosite is shown in Figure 1. The XRD profile of the Fe<sub>2</sub>O<sub>3</sub> powder is shown in Figure 3. Particle size is 42 nm as calculated using Debye Scherrer method [1,2].

Visual appearance of a fired thick film was shown in Figure 2. The film looks good without any crack. The XRD data showed that the thick films had phase of hematite as shown in Figure 3. The SEM data showed that the thick films were porous with relatively small grain size as shown in Figure 4. Electrical resistance of the ceramics decreases following the increase of temperature indicating a semiconducting property. The property is extrinsic property of Fe<sub>2</sub>O<sub>3</sub> [3]. The electrical resistance of the thick films in air was higher than that of the thick films in air containing ethanol gas as shown in Figure 5. Ethanol gas acts as an electron donor for the Fe<sub>2</sub>O<sub>3</sub> ceramics as found in another ceramic that has been applied as an ethanol sensor [4]. This results showed that the thick film ceramics produced here may be applied for ethanol sensor.



Figure 1. Visual appearance of Fe<sub>2</sub>O<sub>3</sub> powder.

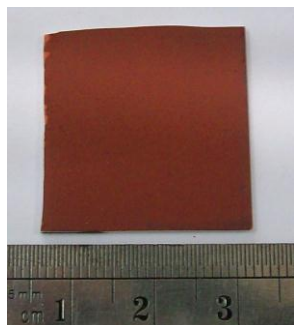


Figure 2. Visual appearance of  $\text{Fe}_2\text{O}_3$  thick film.

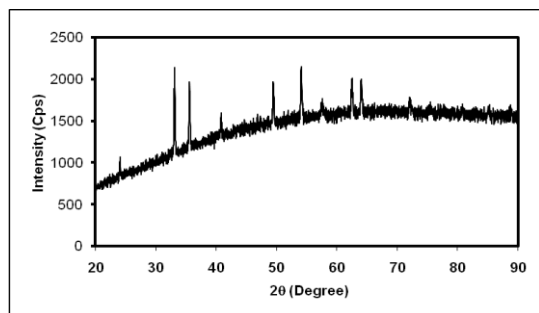


Figure 3. XRD profile of  $\text{Fe}_2\text{O}_3$  thick film.

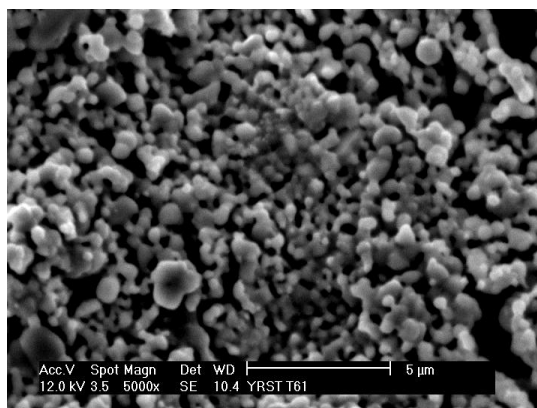


Figure 4. SEM image showing morphology of the  $\text{Fe}_2\text{O}_3$  Thick film.

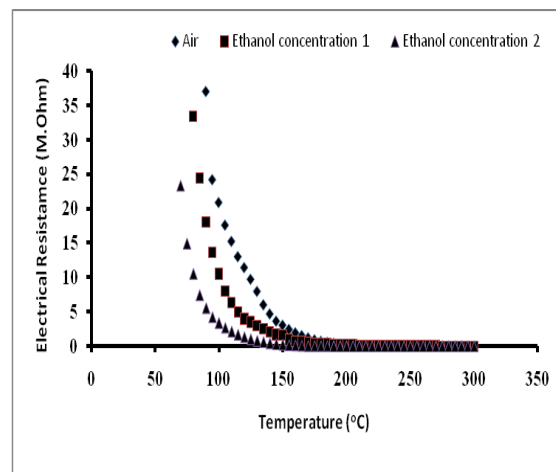


Figure 5. Electrical characteristic of the  $\text{Fe}_2\text{O}_3$  thick film

#### 4. CONCLUSION

Thick film ceramics of  $\text{Fe}_2\text{O}_3$  has been successfully fabricated from local mineral of yarosite. Electrical resistance of the ceramics in ethanol atmosphere was lower than that in air. The ceramics have potentiality to be applied as ethanol sensor.

#### Acknowledgments

The authors wish to acknowledge their deep gratitude to Directorate General of Higher Education (DIKTI), Ministry of National Education of Indonesian Government for financial support under HIBAH BERSAING year 2010.

#### REFERENCES

1. B. Cela, D. A. de Macedo, G. L. de Souza, R. M. do Nascimento, A. E. Martinelli, C. A. Paskocimas, *J. New Mater. Electrochem. Sys.* 12,109-113 (2009).
2. K. S. Rathore, D. Patidar, Y. Janu, N. S. Saxena, K. Sharma, T. P. Sharma, *Chalcogenide Lett.* 5 (6), 105-110 (2008).
3. D. G. Syarif, *Journal of Machine of Trisakti* 9(1), 2007.
4. C. Liewhiran, S. Phanichphant, *Sensors* 7, pp. 650-675, 2007.